Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

N-Benzyl-*N*,4-dimethylbenzenesulfonamide

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Received 27 October 2010; accepted 28 October 2010

Key indicators: single-crystal X-ray study; T = 296 K; mean σ (C–C) = 0.005 Å; R factor = 0.052; wR factor = 0.155; data-to-parameter ratio = 20.5.

The molecule of the title compound, $C_{15}H_{17}NO_2S$, has a C-S-N-C torsion angle of 71.4 (2)°, and the dihedral angle between the benzene rings is 82.83 (16)°. In the crystal, molecules are linked into chains along the *b* axis *via* C-H···O hydrogen bonds. A C-H··· π interaction is also present in the crystal structure.

Related literature

For the pharmacological activities of sulfonamides, see: Maren (1976); Boyd (1988). For our previous studies on derivatives of sulfonamide, see: Khan, Ahmad, Arshad *et al.* (2010); Khan, Ahmad, Sharif *et al.* (2010).



Experimental

Crystal data

$C_{15}H_{17}NO_2S$	b = 8.2632 (7) Å
$M_r = 275.37$	c = 12.0758 (12) Å
Monoclinic, $P2_1/c$	$\beta = 105.902 \ (4)^{\circ}$
a = 15.0386 (16) Å	V = 1443.2 (2) Å ³
a = 15.0386 (16) A	V = 1443.2 (2) A ³

Z = 4Mo $K\alpha$ radiation $\mu = 0.22 \text{ mm}^{-1}$

Data collection

Bruker APEXII CCD diffractometer 13504 measured reflections

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.052$ 174 parameters $wR(F^2) = 0.155$ H-atom parameters constrainedS = 1.01 $\Delta \rho_{max} = 0.18$ e Å $^{-3}$ 3569 reflections $\Delta \rho_{min} = -0.24$ e Å $^{-3}$

Table 1 Hydrogen-bond geometry (Å, °).

Cg2 is the centroid of the C10–C15 phenyl ring.

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
C6-H6···O1 ⁱ	0.93	2.53	3.414 (3)	158
$C9-H9A\cdots Cg2^{ii}$	0.97	2.99	3.721 (3)	133

Symmetry codes: (i) x, y - 1, z; (ii) -x, -y, -z + 1.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).

The authors are grateful to the Higher Education Commission of Pakistan for providing financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5070).

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3569 independent reflections

1580 reflections with $I > 2\sigma(I)$

 $0.28 \times 0.17 \times 0.09 \; \rm mm$

T = 296 K

 $R_{\rm int} = 0.042$

supplementary materials

Acta Cryst. (2010). E66, o3053 [doi:10.1107/S1600536810044156]

N-Benzyl-N,4-dimethylbenzenesulfonamide

I. U. Khan, M. Akkurt, S. Sharif and W. Ahmad

Comment

Sulfonamides have extensively been reported for their wide variety of pharmacological activities such as antibacterial (Maren, 1976) and diuretic (Boyd, 1988). The present structure is continuous to our previous reported derivative of sulfonamide (Khan, Ahmad, Arshad *et al.*, 2010; Khan, Ahmad, Sharif *et al.*, 2010).

In the title molecule (I), (Fig. 1), the molecule has a C4—S1—N1—C9 torsion angle of 71.4 (2)°. The dihedral angle between the sulfonyl benzene ring (C1–C6) and the phenyl ring (C10–C15) is 82.83 (16)°. In the structure, molecules are linked into chains along the *b* axis *via* C—H···O hydrogen bonding (Table 1, Fig. 2). In the structure, there is a C—H··· π interaction (Table 1).

Experimental

A mixture of *N*-benzyl-4-methylbenzenesulfonamide (0.5 g, 2.02 mmol) and sodium hydride (0.2 g, 8.333 mmol) in N,N-dimethylformamide (10 ml) was stirred at room temperature for 30 min followed by the addition of methyl iodide (0.25 ml, 2.02 mmol). After the consumption of reactants (as monitored by TLC), the contents were poured over crushed ice. The precipitated product was isolated, washed, dried and recrystallized from chloroform solution to yield colourless blocks of title compound.

Refinement

All H atoms were positioned geometrically with C—H = 0.93–0.97 Å and treated as riding on their parent atoms, with $U_{iso}(H) = 1.2 \text{ or } 1.5U_{eq}(C)$.

Figures



Fig. 1. The title molecule showing the atomic numbering scheme and 30% probability displacement ellipsoids.



Fig. 2. The packing and hydrogen bonding of the title compound in the unit cell. Hydrogen bonds are shown as dashed lines. For the sake of clarity, the H atoms not involved in the motif have been omitted.

N-Benzyl-N,4-dimethylbenzenesulfonamide

Crystal data

$C_{15}H_{17}NO_2S$
$M_r = 275.37$
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
<i>a</i> = 15.0386 (16) Å
<i>b</i> = 8.2632 (7) Å
<i>c</i> = 12.0758 (12) Å
$\beta = 105.902 \ (4)^{\circ}$
V = 1443.2 (2) Å ³
Z = 4

Data collection

1580 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.042$
$\theta_{\text{max}} = 28.3^{\circ}, \ \theta_{\text{min}} = 1.4^{\circ}$
$h = -19 \rightarrow 20$
$k = -9 \rightarrow 11$
$l = -14 \rightarrow 16$

F(000) = 584 $D_{\rm x} = 1.267 \text{ Mg m}^{-3}$

 $\theta = 2.8-19.5^{\circ}$ $\mu = 0.22 \text{ mm}^{-1}$ T = 296 KBlock, colourless $0.28 \times 0.17 \times 0.09 \text{ mm}$

Mo K α radiation, $\lambda = 0.71073$ Å Cell parameters from 1987 reflections

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.052$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.155$	H-atom parameters constrained
<i>S</i> = 1.01	$w = 1/[\sigma^2(F_o^2) + (0.0492P)^2 + 0.4014P]$ where $P = (F_o^2 + 2F_c^2)/3$
3569 reflections	$(\Delta/\sigma)_{max} < 0.001$
174 parameters	$\Delta \rho_{max} = 0.18 \text{ e } \text{\AA}^{-3}$
0 restraints	$\Delta \rho_{\rm min} = -0.24 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted *R*-factors *wR* and all goodnesses of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The

observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating *-R*-factor-obs *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

	x	У	Ζ	$U_{\rm iso}*/U_{\rm eq}$
S1	0.28540 (5)	0.19774 (9)	0.52072 (8)	0.0817 (3)
01	0.33542 (16)	0.3390 (2)	0.5047 (3)	0.1286 (12)
O2	0.27578 (16)	0.1610 (3)	0.63199 (19)	0.1120 (10)
N1	0.18117 (15)	0.2182 (2)	0.4348 (2)	0.0696 (9)
C1	0.40480 (17)	-0.2407 (4)	0.3876 (2)	0.0683 (10)
C2	0.41527 (19)	-0.0886 (4)	0.3473 (3)	0.0836 (12)
C3	0.38139 (19)	0.0469 (4)	0.3880 (3)	0.0799 (11)
C4	0.33502 (16)	0.0309 (3)	0.4717 (2)	0.0603 (9)
C5	0.32447 (17)	-0.1215 (3)	0.5129 (2)	0.0647 (10)
C6	0.35966 (18)	-0.2545 (3)	0.4714 (2)	0.0685 (10)
C7	0.4415 (2)	-0.3883 (4)	0.3416 (3)	0.1029 (16)
C8	0.1774 (2)	0.2578 (4)	0.3159 (3)	0.1013 (14)
C9	0.11230 (18)	0.0967 (3)	0.4439 (3)	0.0719 (10)
C10	0.01606 (19)	0.1670 (3)	0.4037 (2)	0.0644 (10)
C11	-0.0138 (2)	0.2751 (4)	0.4709 (3)	0.0804 (12)
C12	-0.1020 (3)	0.3421 (4)	0.4332 (4)	0.0991 (16)
C13	-0.1597 (2)	0.2996 (5)	0.3276 (5)	0.1090 (18)
C14	-0.1292 (3)	0.1911 (5)	0.2621 (4)	0.1128 (17)
C15	-0.0424 (2)	0.1256 (4)	0.2990 (3)	0.0876 (12)
H2	0.44620	-0.07680	0.29080	0.1000*
Н3	0.38960	0.14860	0.35930	0.0960*
H5	0.29330	-0.13430	0.56900	0.0780*
Н6	0.35270	-0.35620	0.50080	0.0820*
H7A	0.47420	-0.45520	0.40460	0.1540*
H7B	0.48270	-0.35560	0.29750	0.1540*
H7C	0.39090	-0.44830	0.29330	0.1540*
H8A	0.19200	0.16320	0.27810	0.1520*
H8B	0.22130	0.34170	0.31490	0.1520*
H8C	0.11630	0.29440	0.27650	0.1520*
H9A	0.12430	0.06140	0.52320	0.0860*
H9B	0.11700	0.00330	0.39720	0.0860*
H11	0.02490	0.30430	0.54230	0.0970*
H12	-0.12190	0.41580	0.47950	0.1190*
H13	-0.21850	0.34440	0.30170	0.1310*
H14	-0.16780	0.16060	0.19100	0.1350*
H15	-0.02290	0.05200	0.25230	0.1050*

|--|

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U ³³	U^{12}	U^{13}	U^{23}
S1	0.0745 (6)	0.0657 (5)	0.1035 (7)	-0.0062 (4)	0.0218 (4)	-0.0202 (4)

supplementary materials

O1	0.1010 (18)	0.0631 (13)	0.222 (3)	-0.0296 (12)	0.0449 (18)	-0.0325 (16)
O2	0.1209 (19)	0.139 (2)	0.0748 (15)	0.0202 (15)	0.0246 (13)	-0.0337 (14)
N1	0.0700 (15)	0.0538 (13)	0.0903 (17)	0.0006 (11)	0.0308 (13)	0.0105 (12)
C1	0.0508 (16)	0.081 (2)	0.0685 (18)	0.0036 (13)	0.0088 (14)	-0.0079 (15)
C2	0.072 (2)	0.102 (2)	0.089 (2)	-0.0027 (17)	0.0428 (17)	0.0009 (19)
C3	0.075 (2)	0.0688 (19)	0.105 (2)	-0.0117 (15)	0.0399 (18)	0.0143 (17)
C4	0.0522 (15)	0.0594 (16)	0.0693 (17)	-0.0067 (11)	0.0168 (13)	-0.0028 (13)
C5	0.0607 (16)	0.0711 (18)	0.0654 (17)	-0.0030 (13)	0.0223 (13)	0.0058 (14)
C6	0.0670 (17)	0.0595 (16)	0.0749 (19)	0.0011 (13)	0.0127 (15)	0.0067 (14)
C7	0.088 (2)	0.109 (3)	0.106 (3)	0.0236 (19)	0.017 (2)	-0.027 (2)
C8	0.100 (2)	0.098 (2)	0.115 (3)	0.0158 (18)	0.045 (2)	0.051 (2)
C9	0.0745 (19)	0.0558 (15)	0.089 (2)	-0.0018 (13)	0.0286 (16)	0.0104 (14)
C10	0.0669 (18)	0.0530 (15)	0.0741 (19)	-0.0062 (13)	0.0209 (15)	0.0130 (14)
C11	0.073 (2)	0.081 (2)	0.086 (2)	-0.0004 (16)	0.0199 (17)	0.0049 (18)
C12	0.086 (3)	0.086 (2)	0.138 (3)	0.0105 (19)	0.052 (3)	0.022 (2)
C13	0.063 (2)	0.104 (3)	0.153 (4)	0.003 (2)	0.018 (3)	0.065 (3)
C14	0.089 (3)	0.122 (3)	0.110 (3)	-0.017 (2)	-0.002 (2)	0.031 (3)
C15	0.086 (2)	0.085 (2)	0.087 (2)	-0.0165 (18)	0.0154 (19)	0.0012 (18)

Geometric parameters (Å, °)

S1—O1	1.430 (2)	C14—C15	1.370 (6)
S1—O2	1.423 (2)	С2—Н2	0.9300
S1—N1	1.634 (2)	С3—Н3	0.9300
S1—C4	1.746 (3)	С5—Н5	0.9300
N1—C8	1.459 (4)	С6—Н6	0.9300
N1—C9	1.468 (3)	С7—Н7А	0.9600
C1—C2	1.372 (5)	С7—Н7В	0.9600
C1—C6	1.369 (4)	С7—Н7С	0.9600
C1—C7	1.506 (5)	C8—H8A	0.9600
C2—C3	1.376 (5)	C8—H8B	0.9600
C3—C4	1.383 (4)	C8—H8C	0.9600
C4—C5	1.379 (3)	С9—Н9А	0.9700
C5—C6	1.373 (4)	С9—Н9В	0.9700
C9—C10	1.511 (4)	C11—H11	0.9300
C10-C11	1.363 (4)	C12—H12	0.9300
C10—C15	1.371 (4)	C13—H13	0.9300
C11—C12	1.394 (6)	C14—H14	0.9300
C12—C13	1.377 (7)	C15—H15	0.9300
C13—C14	1.356 (6)		
O1···C6 ⁱ	3.414 (3)	H2···H7B ^x	2.5000
O1···C7 ⁱⁱ	3.384 (4)	H3…O1	2.6500
O2…C8 ⁱⁱⁱ	3.061 (4)	Н5…О2	2.5900
O1…H6 ⁱ	2.5300	H5…C13 ^{ix}	2.9700
O1…H8B	2.4600	H6…O1 ^v	2.5300
O1…H3	2.6500	Н6…Н7А	2.5500
O2…H5	2.5900	H7A…H6	2.5500
O2…H9A	2.4400	H7A…H7A ^{xi}	2.3400

O2…H7C ^{iv}	2.8300	H7B…H2	2.3600
O2…H8A ⁱⁱⁱ	2.8300	H7B…H2 ^{viii}	2.5000
O2…H8B ⁱⁱⁱ	2.5600	H7C…O2 ^{xii}	2.8300
C3···C8	3.427 (5)	Н8А…С3	2.9500
C5…C9	3.559 (4)	H8A…C4	2.9200
C6…O1 ^v	3.414 (3)	H8A…H9B	2.4400
C7…O1 ⁱⁱ	3.384 (4)	H8A…O2 ^{vi}	2.8300
C8…C15	3.434 (5)	H8B…O1	2.4600
C8…C3	3.427 (5)	H8B…O2 ^{vi}	2.5600
C8O2 ^{vi}	3.061 (4)	H8C…C10	2.6500
C9···C5	3,559 (4)	H8C…C15	2.8400
C15…C8	3.434 (5)	H8C···C15 ^{xiii}	3.0000
C2H13 ^{vii}	3,0600	H8CH15 ^{xiii}	2 5200
C2 III5	2 9500	H9AO2	2.3200
C4···H8A	2.9200	H9A…H11	2.5500
C6H14 ^{vii}	3 0900	H9B…H8A	2.3300
	3,0500	H9BH15	2 3700
C/H2	2,6500	H11H0A	2.5700
	2.0300		2.5500
	2.9700		2,0000
	2.8400		2.2700
C15H8C**	3.0000	н15…н9В	2.3700
H2…H7B	2.3600	H15···H8C ^{vn}	2.5200
H2…C7 ^x	3.0500		
O1—S1—O2	119.68 (18)	C4—C5—H5	120.00
O1—S1—N1	106.15 (14)	С6—С5—Н5	120.00
O1—S1—C4	108.00 (14)	С1—С6—Н6	119.00
O2—S1—N1	107.07 (14)	С5—С6—Н6	119.00
O2—S1—C4	108.37 (13)	C1—C7—H7A	110.00
N1—S1—C4	106.92 (11)	С1—С7—Н7В	110.00
S1—N1—C8	114.8 (2)	С1—С7—Н7С	109.00
S1—N1—C9	117.09 (18)	H7A—C7—H7B	109.00
C8—N1—C9	112.9 (2)	Н7А—С7—Н7С	109.00
C2—C1—C6	117.8 (3)	H7B—C7—H7C	109.00
C2—C1—C7	121.4 (3)	N1—C8—H8A	109.00
C6—C1—C7	120.7 (3)	N1—C8—H8B	109.00
C1—C2—C3	121.9 (3)	N1—C8—H8C	109.00
C2—C3—C4	119.6 (3)	H8A—C8—H8B	110.00
S1—C4—C3	121.4 (2)	H8A—C8—H8C	109.00
S1-C4-C5	119 77 (19)	H8B—C8—H8C	110.00
C3—C4—C5	118.8 (2)	N1—C9—H9A	110.00
C4—C5—C6	120 4 (2)	N1—C9—H9B	110.00
C1 - C6 - C5	121.5 (2)	C10-C9-H9A	110.00
N1 - C9 - C10	121.3(2) 1103(2)	C10-C9-H9B	110.00
C_{0} C_{10} C_{11}	10.5(2) 120.2(3)		108.00
$C_{0} = C_{10} = C_{11}$	120.2(3) 121 1(3)	10A - 0 - 117B	120.00
07-010-015	121.1 (3)		120.00

supplementary materials

C11—C10—C15	118.7 (3)	C12—C11—H11	120.00
C10-C11-C12	120.3 (3)	C11—C12—H12	120.00
C11—C12—C13	120.2 (4)	C13—C12—H12	120.00
C12—C13—C14	118.7 (4)	С12—С13—Н13	121.00
C13—C14—C15	121.1 (4)	C14—C13—H13	121.00
C10-C15-C14	121.0 (3)	C13—C14—H14	120.00
C1—C2—H2	119.00	C15—C14—H14	119.00
С3—С2—Н2	119.00	C10-C15-H15	120.00
С2—С3—Н3	120.00	C14—C15—H15	119.00
С4—С3—Н3	120.00		
O1—S1—N1—C8	50.6 (2)	C7—C1—C2—C3	179.5 (3)
O2—S1—N1—C8	179.5 (2)	C1—C2—C3—C4	-0.2 (5)
C4—S1—N1—C8	-64.5 (2)	C2—C3—C4—C5	0.4 (4)
O1—S1—N1—C9	-173.5 (2)	C2—C3—C4—S1	-176.6 (2)
O2—S1—N1—C9	-44.5 (2)	S1—C4—C5—C6	177.2 (2)
C4—S1—N1—C9	71.4 (2)	C3—C4—C5—C6	0.1 (4)
O1—S1—C4—C3	-26.5 (3)	C4—C5—C6—C1	-0.9 (4)
O2—S1—C4—C3	-157.5 (2)	N1-C9-C10-C11	-74.0 (3)
N1—S1—C4—C3	87.4 (2)	N1-C9-C10-C15	105.2 (3)
O1—S1—C4—C5	156.6 (2)	C9-C10-C11-C12	179.0 (3)
O2—S1—C4—C5	25.5 (2)	C15-C10-C11-C12	-0.2 (5)
N1—S1—C4—C5	-89.6 (2)	C9-C10-C15-C14	-179.2 (3)
C8—N1—C9—C10	-68.8 (3)	C11-C10-C15-C14	0.0 (5)
S1—N1—C9—C10	154.48 (19)	C10-C11-C12-C13	0.0 (6)
C6—C1—C2—C3	-0.5 (4)	C11-C12-C13-C14	0.5 (6)
C7—C1—C6—C5	-178.9 (3)	C12-C13-C14-C15	-0.7 (6)
C2-C1-C6-C5	1.1 (4)	C13-C14-C15-C10	0.4 (6)

Symmetry codes: (i) *x*, *y*+1, *z*; (ii) -*x*+1, -*y*, -*z*+1; (iii) *x*, -*y*+1/2, *z*+1/2; (iv) *x*, -*y*-1/2, *z*+1/2; (v) *x*, *y*-1, *z*; (vi) *x*, -*y*+1/2, *z*-1/2; (vii) -*x*, *y*-1/2, -*z*+1/2; (viii) -*x*+1, *y*-1/2, -*z*+1/2; (ix) -*x*, -*y*, -*z*+1; (x) -*x*+1, *y*+1/2, -*z*+1/2; (x) -*x*+1, -*y*-1, -*z*+1; (xii) *x*, -*y*-1/2, *z*-1/2; (xiii) -*x*, *y*+1/2, -*z*+1/2.

Hydrogen-bond geometry (Å, °)

Cg2 is the centroid of the C10–C15 p	ohenyl ring.			
D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A
С5—Н5…О2	0.93	2.59	2.938 (3)	103
C6—H6…O1 ^v	0.93	2.53	3.414 (3)	158
C8—H8B…O1	0.96	2.46	2.886 (5)	107
C8—H8B···O2 ^{vi}	0.96	2.56	3.061 (4)	113
С9—Н9А…О2	0.97	2.44	2.903 (4)	109
C9—H9A····Cg2 ^{ix}	0.97	2.99	3.721 (3)	133

Symmetry codes: (v) x, y-1, z; (vi) x, -y+1/2, z-1/2; (ix) -x, -y, -z+1.



Fig. 1

Fig. 2

server and the server